

# Application News

LC/MS

## No. C225

## Direct Injection Analysis of Organofluorine Compounds (PFAS) by Triple-Quadrupole LC/MS/MS

Organofluorine compounds such as perfluorooctanoic acid (PFOA) and perfluorooctanesulfonic acid (PFOS) (Per/polyfluoroalkyl compound: PFAS) have superior functionality of water and oil repellency and are widely used as coating agents for daily necessities. However, PFAS is known to remain in the blood and is reported to be a toxic compound. Moreover, PFAS is chemically stable and highly persistent. PFOS, a typical organofluorine compound, is listed in Annex B of the Stockholm Convention on Persistent Organic Pollutants (Restriction), and its manufacture and use are internationally restricted. It is designated as a Class I Specified Chemical Substance under the Chemical Substances Control Law in Japan, and its manufacture and use are basically prohibited, with some exceptions.

Analyses of many types of organofluorine compounds have generally been reported after solid-phase extraction/concentration pretreatment, while simplification of pretreatment is required. In this news, PFOA, PFOS, and PFAS including related substances were analyzed by triplequadrupole LC/MS/MS without the concentration procedure.

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## Analysis of Reference Standards

PFOA and PFOS Reference Standards were dissolved and mixed in the solvent, water/methanol = 7/3 (v/v), to prepare a standard sample for the calibration curve. The analytical conditions are shown in Table 1.

In addition, PFOA from the on-line degassing unit and/or flow channel may be concentrated in the analytical column and eluted at the same retention time as the PFOA injected as the analytical sample. To separate background PFOA in the HPLC system from PFOA in the injected sample, a delay column was mounted between the gradient mixer and the autosampler (Figure 1).

This method of delaying the elution of impurities in the system from the target components in the sample is called the Impurity Delay method. By installing a delay column, the elution of PFOA in the system is delayed, and it is possible to analyze PFOA in the sample, as it is isolated.

Table 1 Analytical conditions				
[HPLC conditions] (Nexera <sup>™</sup> Series)				
Column	: Shim-pack Velox <sup>™</sup> SP-C18			
(150 mm L $\times$ 2.1 mm l.D., 2.7 µm, PN: 227-32003-04)				
Delay column : Shim-pack <sup>™</sup> XR-ODS II				
(75 mm >	< 2.0 mm l.D., 3 μm, PN: 228-41623-91)			
Mobile phases	: A) 20 mmol/L Ammonium Acetate in $H_2O$			
	B) Methanol			
Gradient program	: B 60% (0.00 min) - B85% (25.00 - 30.00 min)			
	- 60% (30.01 - 34.00 min)			
Flow rate	: 0.25 mL/min			
Column temp.	: 40 °C			
Injection volume	: 50 μL			
[MS conditions] (LCMS	<sup>TM</sup> -8060)			
lonization	: ESI (Negative mode)			
Probe voltage	:-1 kV			
Mode	: MRM			
Nebulizing gas flow	: 3 L/min			
Drying gas flow	: 5 L/min			
Heating gas flow	: 15 L/min			
DL temp.	: 200 °C			
Heat block temp.	: 300 °C			
Interface temp.	: 300 °C			

[MS/MS paramet	ers]	
Compound	MRM transition (m/z)	Collision energy (V)
PFOA	412.90>369.05	10.0
	412.90>169.10	18.0
PFOS	498.90>79.95	55.0
	498.90>98.95	40.0

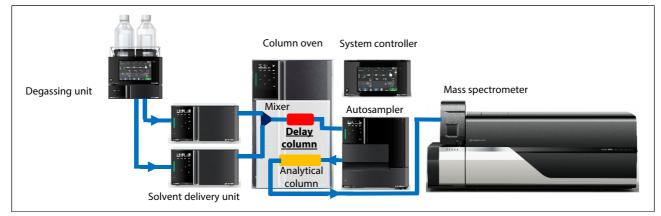


Figure 1 Impurity delay method

## Results of the Standard Analysis

Calibration curves from 1 to 100 ng/L are shown in Figure 2. These calibration curves for both PFOA and PFOS indicate good linearity with R2 > 0.999.

Table 2 shows the accuracy (%) at each calibration point. The accuracy of the calibration curve was 91.4 to 101.5%, which is a good result. Figure 3 shows each chromatogram of the 1 ng/L standard sample for PFOA and PFOS, respectively. The repeatability (n = 3) of the standard sample at this concentration was 5.8% for PFOA and 4.9% for PFOS, indicating good reproducibility.

Table 2 Accuracy at the calibration points (%)

	PFOA	PFOS
1 ng/L	93.5	92.7
2 ng/L	101.4	91.4
5 ng/L	94.6	95.4
10 ng/L	101.4	101.4
20 ng/L	102.2	101.5
50 ng/L	99.3	100.3
100 ng/L	100.1	99.9

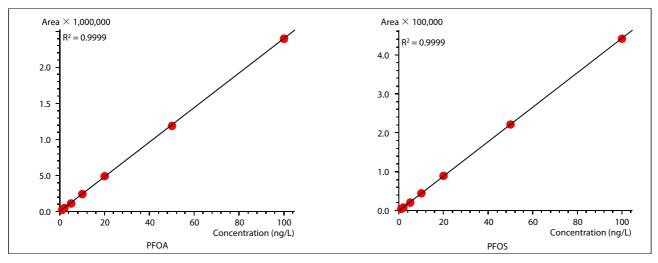


Figure 2 Calibration curves for PFOA and PFOS

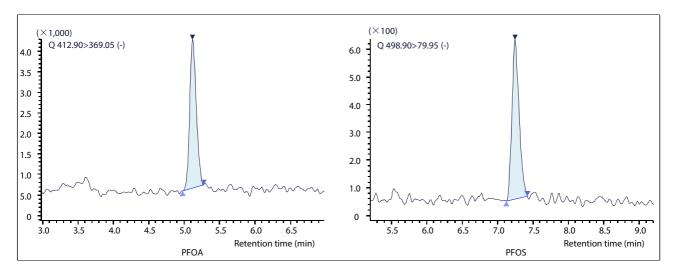


Figure 3 Chromatogram of POA and PFOS at 1 ng/L standard solution

## Analysis of PFOA / PFOS Analogues

The reinforcement of PFAS regulations is advancing worldwide, and, at the same time, Method 537 of the U.S. Environmental Protection Agency (EPA) and ASTM D 7979 define analytical methods for PFOA and PFOS analogues.

Twenty-nine components (Table 3) including PFOA and PFOS were analyzed by the direct injection method using the Nexera MX system shown in Figure 4. This analysis was conducted under the conditions shown in Table 4, and calibration curves were prepared based on the internal standard method.

#### Table 3 List of compounds to be analyzed

	Compounds	Composition formula	Monoisotopic mass	
1	4:2 FTS	$C_6H_5F_9SO_3$	327.9816	
2	6:2 FTS	$C_8H_5F_{13}SO_3$	427.9752	
3	8:2 FTS	$C_{10}H_5F_{17}SO_3$	527.9688	
4	10:2 FTS	$C_{12}H_5F_{21}SO_3$	627.9624	
5	N-EtFOSA	$C_{10}H_{6}F_{17}NSO_{2}$	526.9848	
6	N-EtFOSAA	$C_{12}H_8F_{17}NSO_4$	584.9903	
7	N-EtFOSE	$C_{12}H_{10}F_{17}NSO_3$	571.0110	
8	FOSA	$C_8H_2F_{17}NSO_2$	498.9535	
9	N-MeFOSAA	$C_{11}H_{6}F_{17}NSO_{4}$	570.9746	
10	N-MeFOSE	$C_{11}H_8F_{17}NSO_3$	556.9953	
11	N-MeFOSA	$C_9H_4F_{17}NSO_2$	512.9691	
12	PFTrDA	$C_{13}HF_{25}O_2$	663.9577	
13	PFBA	C <sub>4</sub> HF <sub>7</sub> O <sub>2</sub>	213.9865	
14	PFBS	C <sub>4</sub> F <sub>9</sub> SO <sub>3</sub> H	299.9503	
15	PFDA	$C_{10}F_{19}O_2H$	513.9673	
16	PFDoDA	$C_{12}F_{23}O_2H$	613.9609	
17	PFHpA	$C_7F_{13}O_2H$	363.9769	
18	PFHxA	$C_6F_{11}O_2H$	313.9801	
19	PFHxS	$C_6F_{13}SO_3H$	399.9439	
20	PFPeS	$C_5F_{11}SO_3H$	349.9471	
21	PFNA	$C_9F_{17}O_2H$	463.9705	
22	PFOA	$C_8F_{15}O_2H$	413.9737	
23	PFDS	$C_{10}F_{21}SO_3H$	599.9311	
24	PFHpS	C <sub>7</sub> F <sub>15</sub> SO <sub>3</sub> H	449.9407	
25	PFOS	C <sub>8</sub> F <sub>17</sub> SO <sub>3</sub> H	499.9375	
26	PFPeA	$C_5HF_9O_2$	263.9833	
27	PFTeDA	C <sub>14</sub> HF <sub>27</sub> O <sub>2</sub>	713.9545	
28	PFUnDA	$C_{11}HF_{21}O_2$	563.9641	
29	PFHxDA	$C_{16}HF_{31}O_2$	813.9482	

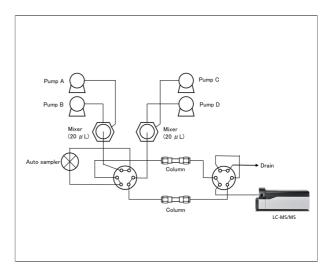


Figure 4 Flow diagram (High-throughput analysis)

## Table 4 Analytical conditions

	,			
[HPLC conditions] (Nexera Series)				
Column	: Shim-pack ODS 🎞			
(50 mm x 2.0 mm l.D., 1.6 μm, P/N: 228-59922-91)				
Delay column	: Shim-pack XR-ODS			
(30 mm x 3	.0 mm I.D., 2.2 μm, P/N: 228-41606-91)			
Mobile phases	: A) 5mM Ammonium Acetate / 0.05% Acetic Acid in H <sub>2</sub> O B) Methanol			
Gradient program	: B 50%(0.00 min) – B100%(4.60-5.50 min)			
Flow rate	: 0.4 mL/min			
Column temp.	: 40 °C			
Injection volume	: 40 μL			
[MS conditions] (LCMS-8	<u>050)</u>			
lonization	: ESI (Negative mode)			
Probe voltage	:-3 kV			
Mode	: MRM			
Nebulizing gas flow	: 2.6 L/min			
Drying gas flow	: 7 L/min			
Heating gas flow	: 9 L/min			
DL temp.	: 180 °C			
Heat block temp.	: 250 °C			
Interface temp.	: 350 °C			

## Results of the Analogue Analysis

The TIC chromatogram obtained by measuring the standard mixture solution is shown in Figure 5. An analytical cycle is 5.5 minutes. The ranges of calibration curve and contribution ratios for each compound are shown in Table 5.

A good calibration curve with a contribution ratio ( $R^2$ ) > 0.99 was obtained for each compound, and it was shown that the simultaneous analysis of analogues including PFOA and PFOS can be performed with high throughput.

## Conclusions

- Samples of 1 ng/L could be analyzed for PFOA and PFOS without concentration.
- A total of 29 components including PFOA, PFOS, and their analogues could be simultaneously analyzed in just 5.5 minutes.
- The major PFAS can be determined without the solid-phase extraction or concentration steps described in EPA METHOD 537.1.

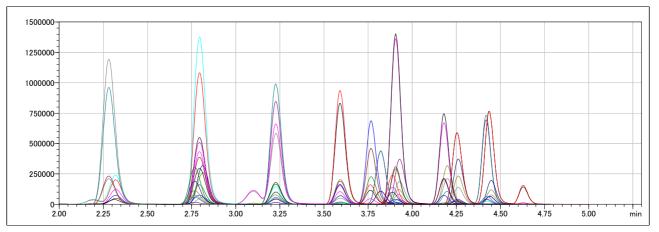


Figure 5 TIC Chromatogram of the standard mixture (each concentration;  $1 - 25 \,\mu g/L$ )

	Compounds	Calibration range (µg/L)	Contribution ratio (R <sup>2</sup> )		Compounds	Calibration range (µg/L)	Contribution ratio (R <sup>2</sup> )
1	4:2 FTS	0.0934 – 2.3350	0.9936	16	PFDoDA	0.0100 - 5.0000	0.9990
2	6:2 FTS	0.0095 – 2.3700	0.9995	17	PFHpA	0.0100 - 5.0000	0.9999
3	8:2 FTS	0.0096 - 2.3950	0.9948	18	PFHxA	0.0100 - 5.0000	0.9999
4	10:2 FTS	0.0096 - 4.8200	0.9969	19	PFHxS	0.0095 - 4.7300	0.9992
5	N-EtFOSA	0.0250 – 12.5000	0.9984	20	PFPeS	0.0094 - 4.6900	0.9992
6	N-EtFOSAA	0.0400 - 5.0000	0.9976	21	PFNA	0.0100 - 5.0000	0.9999
7	N-EtFOSE	0.0250 - 12.5000	0.9997	22	PFOA	0.0100 - 5.0000	0.9999
8	FOSA	0.0100 - 5.0000	0.9998	23	PFDS	0.0096 - 0.9640	0.9991
9	N-MeFOSAA	0.0400 - 5.0000	0.9991	24	PFHpS	0.0095 – 2.3800	0.9990
10	N-MeFOSE	0.0250 - 12.5000	0.9986	25	PFOS	0.0093 - 9.2800	0.9996
11	N-MeFOSA	0.0250 - 12.5000	0.9972	26	PFPeA	0.0100 - 5.0000	0.9999
12	PFTrDA	0.0100 - 5.0000	0.9960	27	PFTeDA	0.0250 - 12.5000	0.9999
13	PFBA	0.0500 - 25.0000	0.9991	28	PFUnDA	0.0100 - 5.0000	0.9991
14	PFBS	0.0088 - 4.4200	0.9990	29	PFHxDA	0.0100 - 2.5000	0.9990
15	PFDA	0.0100 - 5.0000	0.9994				

#### Table 5 Calibration range and contribution ratio (R<sup>2</sup>) for each target compound

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